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CHAETOGLOBOSINS G AND J, CYTOTOXIC INDOL-3-YL[13]-

CYTOCHALASANS FROM Chaetomium globosum

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CYTOCHALASINS<sup>1)</sup> are now attracting attention as a group of mycotoxins and a tool for cell biology.<sup>2)</sup> The chemistry of chaetoglobosins A - F, new members of cytochalasins bearing an indol-3-yl group from *Chaetomium globosum*, was reported in the previous communications.<sup>3-5)</sup> The stereochemistry adopted in the preceding papers<sup>4,5)</sup> for A - D (<u>1</u> - <u>4</u>) was later proved to express the absolute configurations of these compounds.<sup>6)</sup> This communication concerns further evidence for the structures of chaetoglobosins E and F and characterization of two new congeners from the same source.

The structures proposed for E and  $F^{5}$  were based on the comparison of their physical properties, especially of <sup>1</sup>H-NMR data, to those of A - D, whose structures had been established by X-ray analyses<sup>4,6</sup>) and correlation reactions. Now chemical proof of the structures has been obtained as follows: Treatment of chaetoglobosin F in boiling acetic acid forms chaetoglobosin E in a good yield. Oxidation of  $\alpha$ -ketol group in F with Bi<sub>2</sub>O<sub>3</sub> in acetic acid to  $\alpha$ -diketone afforded chaetoglobosin C (3)<sup>5,7</sup>) along with an isomer of B (chaetoglobosin G, 7, vide infra). Thus the compounds were correlated each other.<sup>8</sup>) The stereochemistry of C<sub>20</sub> in E and F was suggested to be (S) from following facts. The conformations of the 13-membered rings in chaeto-globosins A and C in the crystalline state revealed by X-ray analyses (Chart 1)<sup>4,6,7</sup> are indicated as being retained in CDCl<sub>3</sub> or C<sub>6</sub>D<sub>5</sub>N solution by the precise examinations of the chemical shifts and the coupling constants in <sup>1</sup>H-NMR.<sup>8</sup>) The examination of the <sup>3</sup>H-NMR data of E and F and the acetates revealed that the l3-membered ring adopts nearly the same conformation as A - D.

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Since nuclear Overhauser effects were observed between the  $C_{17}$ -olefinic protons with  $C_{20}$ -carbinyl protons both in E and F (Chart 1), the stereochemistry was assigned as shown in the formulae (5) and (6).<sup>8)</sup>

The new chaetoglobosin, named G, colorless leaflets of mp 251-253° from MeOH,  $[\alpha]_{D}$  + 89° (MeOH),  $\lambda_{max}^{EtOH}$  222, 275, 282, 291 nm (log  $\varepsilon$  4.51, 3.79, 3.79, 3.73),  $\nu_{max}^{KBr}$  3455, 3300, 1713, 1693, 1646, 1623, 987, 948, 741 cm<sup>-1</sup>, has the same molecular formula as chaetoglobosins A - D,  $C_{32}H_{36}O_5N_2$  (M<sup>+</sup> 528.269 m/e, calcd., 528.262). It forms a monoacetate. The <sup>1</sup>H-NMR spectra of G and the acetate indicated that the perhydroisoindolone part of the molecule is same as chaetoglobosins B (2) and E (5), while the 13-membered ring portion is same as chaetoglobosin C (3).<sup>8</sup> Indeed the cleavage of the epoxide ring of chaetoglobosin C (3) with HOAc, under the same conditions as B (2) was formed from A (1), afforded the new congener, while treatment of C (3) with HOAc-H<sub>2</sub>SO<sub>4</sub> gave the acetate of G. On the other hand treatment of B (2) with Et<sub>3</sub>N-pyridine, under the same conditions as C (3) was produced from A (1), afforded the new congener. These reactions clearly demonstrated that the structure of chaetoglobosin G must be the formula (7).



The other new member, chaetoglobosin J, pale yellow prisms of mp 149-151° from benzene,  $\lambda_{max}^{Et0H}$  224, 270, 280, 290 nm (log  $\varepsilon$  4.68, 3.86, 3.86, 3.78),  $\nu_{max}^{KBr}$  3412, 3273, 1683, 1639, 1612, 980, 975, 925, 750 cm<sup>-1</sup>, has a molecular formula,  $C_{32}H_{36}O_4N_2$  (M<sup>+</sup> 512.258 m/e, calcd. 512.250), which corresponds to a deoxygenated compound of chaetoglobosins A - D and suggests a similarity to zygosporin G<sup>9</sup> and proxiphomin<sup>10</sup> in the case of 10-phenylcytochalasans. It forms a monoacetate. Indeed <sup>1</sup>H-NMR spectra of J and the acetate revealed that the 13-membered ring is same as A (<u>1</u>), B (<u>2</u>), and D (<u>4</u>) but the perhydroisoindolone part was assigned as shown in Chart 2. Direct proof of the structure was obtained by the deoxygenation reaction of chaetoglobosin A monoacetate with WCl<sub>6</sub>-BuLi<sup>11</sup> to give the acetate of chaetoglobosin J and the structure



with WCl<sub>6</sub>-BuLi<sup>11)</sup> to give the acetate of chaetoglobosin J and the structure ( $\underline{8}$ ) was established for the new congener.

Although chaetoglobosins G and J exhibited nearly the same cytotoxicity to cultured HeLa cells at ca 3  $\mu$ g/ml, it is noteworthy that, unlike other members of the group, chaetoglobosin J does not form multinucleated cells.<sup>12</sup>)

## References

- 1) M. Binder, Ch. Tamm: Angew. Chem., Intern. Ed., 12, 370 (1973).
- 2) S. B. Carter: Endeavour, 113, 77 (1972): S. Natori: Proc. of UJNR Conference on Mycotoxins to Human and Animal Health, College Park, Md., U. S. A., 1976, in the press.
- 3) S. Sekita, K. Yoshihira, S. Natori, H. Kuwano: Tetrahedron Letters, 1973, 2109.
- J. V. Silverton, T. Akiyama, C. Kabuto, S. Sekita, K. Yoshihira, S. Natori: Tetrahedron Letters, <u>1976</u>, 1349.
- 5) S. Sekita, K. Yoshihira, S. Natori, H. Kuwano: Tetrahedron Letters, 1976, 1351.
- 6) J. V. Silverton, C. Kabuto, T. Akiyama: Submitted to Acta Crystallographica (1977).
- 7) J. P. Springer, J. Clardy, J. M. Wells, R. J. Cole, J. W. Kirksey, R. D. MacFarlane, D. F. Torgerson: Tetrahedron Letters, <u>1976</u>, 1355.
- 8) S. Sekita, K. Yoshihira, S. Natori, H. Kuwano, M. Umeda: 20th Symposium on the Chemistry of Natural Products, Sendai, Japan, 1976, Symposium Papers, p. 396.
- 9) H. Minato, T. Katayama: J. Chem. Soc., C, 1970, 45.
- 10) M. Binder, Ch. Tamm: Helv. Chim. Acta, 56, 2387 (1973).
- K. B. Sharpless, M. A. Umbreit, M. T. Nieh, T. C. Flood: J. Am. Chem. Soc., 94, 6538 (1972);
  K. Yamada, H. Tatematsu, Y. Hirata, M. Haga, I. Hirono: Chemistry Letters, 1976, 1123.
- 12) M. Umeda: Private communication.